SURFACE FINISHING IN THE ANALYSIS OF ALUMINIUM ALLOYS BY X-RAY FLUORESCENCE SPECTROMETRY

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ABSTRACT

The determination of the surface finishes of maximum roughness permissible for solid aluminium alloy specimens intended to be analyzed for silicon, iron, copper, magnesium, zinc, manganese, titanium, chromium and nickel by X-ray fluorescence on the Picker Nuclear Spectrodiffractometer.

The determination of the value of "K" for the optical geometry associated with the above instrument.

The determination of the applicability of engine lathe machining as a method of surface finishing allowing the determination quantitatively of the elements above, while meeting the requirements as to speed of finishing demanded by industrial analytical processes.

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1. INTRODUCTION AND THEORY

Solid aluminium alloy samples, secured by methods similar to those used in the sampling of molten aluminium alloys for chemical analysis by emission spectrography, can be analyzed both qualitatively and quantitatively by X-ray fluorescence spectrometry. The method is, of course, subject to the usual limitations as to quantity and element commonly associated with this technique of analysis. For example, elements of atomic number lower than sodium are not determinable with any worthwhile degree of accuracy using equipment that is standard in industrial applications.

The technique has, over recent years, been used as both a substitute for and an adjunct to the analysis of metallic materials by emission spectrography and, because of:

- a) less complicated elemental spectra than the comparable emission spectra,
- b) less restrictions as to the range of quantity determinable,
- c) a generally improved relative error situation

has provided for, in many instances, an improved pattern of qualitative and quantitative results.

In brief, the technique as applied to aluminium alloys involves the irradiation of a properly prepared aluminium alloy solid sample surface with x-rays. The source of these x-rays is an x-ray tube, the anode of which is usually tungsten for the detection of elements of atomic number higher than scandium, or chromium in the case of elements of atomic number lower than titanium, although other tube anode materials are quite commonly encountered.

The primary beam from the anode causes the emission of secondary x-ray radiation by the sample, the pattern of emitted radiation representing a combination resulting from the elements present in the sample matrix. The collimated secondary beam is passed to a diffraction crystal, the nature of which will depend on the wavelengths for the impimging radiation. Subsequently the diffracted radiation is detected and measured, by either a scintillation counter or a proportional flow counter, at a 2-theta angle appropriate to the radiation wavelength involved and the diffraction crystal used. Figure 1 represents the general arrangement.

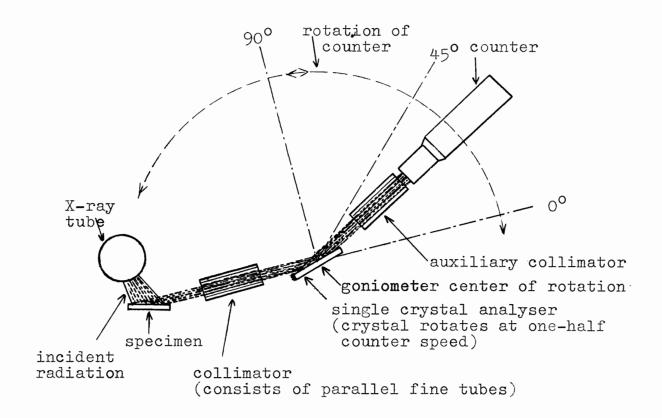


Figure 1.

Basic geometry of X-ray emission spectrograph.

Detection of diffracted radiation at specific 2-theta angles serves to identify the matrix elements involved; the intensity of the radiation measured at each angle by the counter permits quantitative estimation.

2. PURPOSE

The importance of the method of preparation of the surface of solid samples to be analyzed by x-ray fluorescence (XRF) has been stressed by several authors, notably Michaelis and Kilday (1), Flikkema and Schablaske (2), Gidley (3), Bills (4) and (5), Kilday and Michaelis (6), Lihl and Fischhuber (7) and Jenkins and Hurley (8). Basically two aspects relative to surface finish and preparation are of importance. These are:-

- a) the separation distance between the sample surface and the x-ray tube anode
- b) the shielding effect of the surface maxima arising out of the general surface finish.

 An elaboration of these factors follows.

a) SEPARATION DISTANCE

This represents the distance separating the

anode of the X-ray tube and the surface of the sample under examination. The effect of increasing separation will be, of course, to lower the intensity of the primary beam at the sample surface and, consequently, the intensity of any secondary radiation emitted by the sample. In X-ray fluorescence spectrometers the sample is located at a fixed distance from the anode. Surface irregularities produced by the method of surface preparation do, however, result in a difference between the distance of separation of the peaks and valleys of these irregularities from the anode. An investigation by Jenkins and Hurley (8) indicated that, using several

selected wavelengths of secondary emission, no significant change in emitted intensity took place until an increase in separation distance of 50 μ (1 μ = 10 $^{-6}$ meters) was achieved. At a separation distance of 500 μ the decrease in intensity amounted to about 3 percent, and thereafter the decrease in intensity amounted to about 1 percent for each additional 90 μ increase in separation distance. Generally speaking, the effects observed were independent of the emitted wavelength, but were strongly dependent on instrumental geometry, decreasing with

increasing fixed distance from sample surface to anode and increasing with decreasing take-off angle from the sample surface.

Since the surface is unlikely to be worse than an average of 50 μ distance from peak to valley of the irregularities as a commercial practice this effect, for a given instrument, will be relatively unimportant.

b) SHIELDING EFFECT

This effect has been described by Gunn (9). Figure 2 shows the effective depth of penetration of the sample surface by primary x-rays as the dotted lines, while the sample surface is shown as the solid line. A primary beam entering the sample at an angle of ψ_l , penetrates to the effective depth, so that the cross-hatched area of the sample does not contribute in any way to the intensity of the secondary radiation emitted. The effective depth of penetration can be estimated from the take-off angle ψ_{λ} of the spectrometer, and the value of "x", the path length of the emitted radiation to be measured. The value of "x" can be calculated from the standard expression for absorption:-

$$I = I_o e^{-(u_m dx)}$$
 (1)

where \mathbf{u}_{m} is the mass absorption coefficient of the

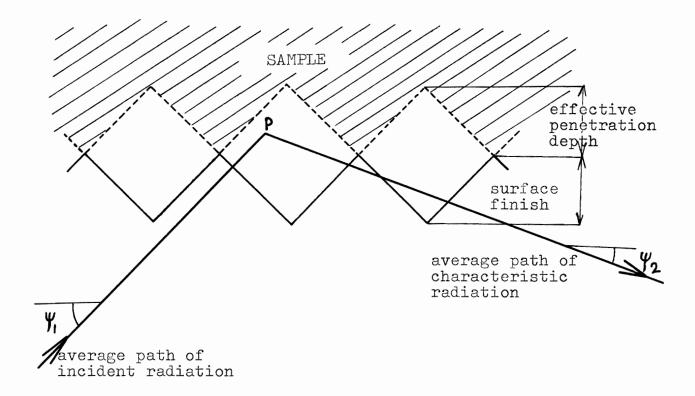


Figure 2.

Effect of shielding due to surface finish.

matrix for the wavelength of the measured radiation, and "d" is the density of the matrix. For a value of 99 percent absorbance we have:-

and:-
$$x = \frac{4.6052}{u_m d}$$
 in cm
$$= \frac{46052}{u_m d}$$
 in microns (μ)

Any radiation arising at the point P, although it lies within the effective depth of penetration, will not be measurable, because of the additional interposed path length of the surface maximum involved.

Obviously, the more severe the surface maxima (rougher surfaces) the greater will be this effect in reducing the measurable intensity of emitted radiation. Just as obviously, the greater the take-off angle, the less this effect for comparable surface maxima. Again, the effect will be somewhat reduced with shorter emitted wavelengths (i.e. longer path lengths).

The work of Jenkins and Hurley (8) showed that it was possible, from data secured relative to surface roughness and emitted intensity of secondary radiation, to correlate path length and surface finish, and to

determine an expression for the estimation of the maximum allowable surface roughness (peak-to-valley distance) for the characteristic wavelength emitted by a given element in a given metallic matrix. Their work, where it covered aluminium alloys, reported only $S_{\rm max}$ (surface roughness maximum) values for aluminium, silicon, iron, copper. This, together with data accumulated on other metallic matrices, permitted the formulation:

$$S_{\text{max}} = K \cdot \left(\frac{46052}{\mu_{\text{m}}d}\right)^{1/3} \quad \text{microns } (\mu) \quad (2)$$

where: S_{max} is the surface finish in terms of the average peak-to-valley separation in microns.

K is a factor dependent on the instrumental geometry. (For a take-off angle of 35°, K was found to be 30).

It will be noted that S_{max} varies as the one-third power of "x" the path length of the emitted radiation to be measured.

Mention should be made of a third factor of some considerable importance in sample surface prepation. This is the property of certain elements, notably silicon and lead, of producing surface smearing effects under standard lathe turning or abrasive

methods of surface finishing. Such surface layers result from the detachment of softer particles of these elements from the matrix, and the smearing of the particles over the surface. While this effect can be minimized by carefully controlled techniques of surface finishing, where it occurs it can cause either or both:

- a) enhancement of the intensity of the emitted radiation for the smeared element,
- b) decrease by absorption of the intensity of emitted radiation for certain lower atomic number matrix elements.

The available methods of surface preparation for aluminium alloys commonly applied are:

- a) lathe turning,
- b) abrasive finishing.

Although many XRF operations are based on the reduction of the sample surface by abrasive finishing, care must be taken to avoid final finishing operations with abrasive papers, cloths or powders involving silicon carbide or aluminium oxide, since these two substances may permit erratic results, particularly in the determination of aluminium (for Al₂O₃ finishes) or silicon (for SiC finishes) in the matrix. Where

final abrasive finishes are involved, a last step with fine diamond abrasive is required to eliminate the possibility of such effects.

The method of abrasive finishing usually involves the use of successively finer grits, culminating with a final operation using 20 micron diamond paste. The method, while excellent, is time-consuming and, in industrial applications where many samples must be handled rapidly, time is usually an extremely important factor. Methods involving lathe turning, if applicable, would provide a much more rapid means of securing the desired surface finish. Care must be taken relative to machine tool shape and sharpness in order to avoid the problems associated with surface smearing.

The purpose of this investigation is therefore to investigate further in depth the work of Jenkins and Hurley (8) relative to:

- a) extending the number of elements in the aluminium alloy matrix for which S_{max} is determined,
- b) since the optical geometry of the Picker Nuclear Instrument differs from that of the Philips equipment used by Jenkins and Hurley (45° take-off instead of 35°, for example), determination of the

value of "K" in equation (2) was required for future reference.

c) determination of the applicability of the method of lathe turning as a rapid means of surface finishing aluminium alloy samples for XRF analysis.

3. METHODS OF DATA AQUISITION

a) Matrix elements to be investigated

The most common elements determined in the aluminium alloy matrix are:

Silicon

Iron

Copper

Magnesium

Manganese

Zinc

Titanium

Chromium

Nickel

It is understood that, for metallurgical reasons, the determination of silicon by XRF is usually restricted to those aluminium alloys where the

silicon content is about 10 % or less. Problems with respect to segregation and resultant lack of compositional homogenity prevent the obtaining of accurate results where the silicon content much exceeds this level.

b) Selection of standard samples for investigation

A selection of standard analyzed samples was made available by the Aluminium Company of Canada. Of these, several were selected relative to the investigation of various matrix elements, and the relationship between surface finish and their emitted radiation intensities. Table 1 shows the selection, compositions and elements investigated.

c) The method of preparing the surface

The Alcan standards were available in discs of the proper diameter to fit the Picker Nuclear sample holder. Only two discs of each alloy were available, but the discs were of a thickness adequate to permit successive surface finishing operations to be carried out.

it was not feasible to cross-section each disc after each surface finishing procedure and X-ray emission intensity measurements in order to determine by measurement the actual surface roughness as an average peak-to-valley value. A series of discs was therefore cut from standard aluminium alloy bar stock, and one of these was subjected to each surface finishing operation. These extra discs were then cross-sectioned and used to determine the surface given by each finishing operation. It was assumed that, in any such comparison, the surface finish or profiles determined for these extra or control discs would hold for the experimental discs.

Each experimental disc, with the control disc for that set, was turned down on a high speed engine lathe at the depths of cut and traverse speeds given in Table 2. The descriptive number under the column headed "SET" is intended to identify each turning operation by the anticipated peak-to-valley value in inches. The lathe tool used had a 60° profile at the carballoy tip and was re-sharpened after each individual set series.

In addition to and after the series of surface finishes produced by lathe turning, each experimental disc was subjected finally to a series of abrasive finishing operations involving successively 180, 240, 320, 400 and 600 grit silicon carbide abrasive cloths. The final finish for each disc was given with a wet polishing wheel using 6 micron diamond paste.

d) Method of determining the surface finish

All of the lathe turned control discs were cross-sectioned at right angles to the finished surface. The cross-sectioned surface was then reduced successively using 180, 240, 320, 400 and 600 grit silicon carbide abrasive cloths. The final finish was given with a wet polishing wheel using 1 micron diamond paste. These cross-sections were then examined at 1000 magnifications under a metallurgical microscope. The projection screen of the microscope was capable of the following measurement limits:-

 $\label{eq:local_problem} \text{l division} \ = \ \text{O.Olo mm} \quad (\text{lo } \mu)$ with an uncertainty of:-

 \pm 0.2 division = \pm 0.002 mm (2 μ)
Each control disc was examined at 10 locations, and

the corresponding peak-tovalley values secured.

Tables 3 to 12 inclusive show the values obtained.

The diamond finished control disc was examined in 10 locations by a Surfmeter tester, and Table 13 shows the results secured.

Table 14 shows the final values of surface finish in microns peak-to-valley separation involved. The optimum value of surface finish for graphing purposes is also shown.

As a matter of interest, the average surface finish produced by various abrasive grit polishes was determined by Surfmeter measurement.

Table 15 shows the associated values in microns. It will be noted that even a finish by 80 grit silicon carbide provides for a surface of lesser roughness than the best lathe turning operation applied in this investigation.

e) Method of securing emission intensity data

The instrumental operating parameters required for the detection of emitted radiation from the matrix elements to be investigated had previously been determined for aluminium alloys by Professor

Dick in association with Mr. A. Fraser and Mr. G. Green. These parameters are shown in table 16.

The intensity of emission for each element involved the following wavelengths.

Element	Radiation	$\lambda(\tilde{\mathbb{A}})$
Silicon	Ka	7.13
Iron	$^{\mathrm{K}}$ a	1.94
Copper	K_{∞}	1.54
Magnesium	Ka	9.89
Manganese	K.	2.10
Zinc	K _{ok}	1.44
Titanium	K	2.75
Chromium	K _A	2.29
Nickel	$^{\mathrm{K}}$	1.66

In order to ensure that the counting level for each series of counts involving the different surface finishes was referred to the same basis, an emission intensity counting standard for each element was used. These counting standards were aluminium alloy discs of fixed and measured surface finish. For each set of surface fihishes for the experimental discs, the counting standard appropriate to the element involved was counted, and a correction factor was subsequently used

to correct the counting intensity for a given element to the same basis for each surface finish. The counting standards used were:

Counting Standard	Element
143-AA	Iron
CA-6195-AA-D	Silicon
6561-CABS	Zinc
143-AA	Copper
143-AA	Chromium
6046-CABS	Titanium
340-AA-D	Magnesium
143-AA	Manganese
6046-CABS	Nickel

The emission intensity for each element, associated with each surface finish, was first corrected for dead time wherever the count rate on the scintillation counter exceeded 30,000 cps. No counts taken on the proportional flow counter exceeded the dead time correction minimum, so that proportional flow counter counting required no dead time corrections. The formulation for dead time correction on the scintillation counter was derived by Professor Dick as:

$$N = \frac{n}{1.02437 - (0.79681 \times 10^{-6} \times n)}$$
 (3)

where:- N = corrected count in cps

n = measured count in cps

Subsequent to the dead time correction, the count intensity was corrected for the average background count. Finally, this latter corrected count was multiplied by the counting standard correction factor. Tables 25 inclusive summarize the data secured for each element and each finishing operation. The final column in each case gives the 3s (3 times standard deviation) counting error in cps for the counting level involved. The second to the last column shows the difference between the counts in cps for each surface finish and the finish immediately It was taken that any difference equal or less rougher. counting error represented the achievement 3s of a constant counting rate and a finished surface finer than which would yield no further change in the intensity of emission.

Figures 3 to 11 show the data from Tables 17 to 25 expressed in graphic form. The straight line

section representing a steady intensity of emission was extrapolated in each case, and the departure point of the curve corresponding to magnitude of difference equal to 3s was taken as the point of critical surface finish or S_{max} . This point is indicated on the extrapolated line by the cut-off "] " position. The following S_{max} values were secured in this way:

Element	Smax	(µ)
Silicon	75	
Iron	265	
Copper	235	
Magnesium	115	
Manganese	200	
Zinc	250	
Titanium	145	
Chromium	135	
Nickel	195	

f) Calculation of mass absorption coeffcients for the various alloy matrices

In order to determine the value of "K" in equation (2)

$$S_{\text{max}} = K \cdot \left(\frac{46052}{\mu_{\text{m}} d}\right)^{1/3} \text{ microns}$$

it is necessary to determine the alloy densities, and the mass absorption coefficients for the various elements relative to the alloy matrices.

The density values were determined by the usual displacement - weight technique. These were found to be sufficiently similar for the alloys involved to allow use of a general density value of 2.7 g/cm³.

Table 26 shows the calculation of the mass absorption coefficient for the relevant elements in the various alloy matrices.

These can be summarized as:

Element Alloy		$\mu_{ ext{m}}$	
Silicon	CA CAOE AA D	2040	
STITCOU	CA-6195-AA-D	2840	
Iron	CA-6195-AA-D	86	
Copper	CA-6195-AA-D	50	
Magnesium	CA-79-S-AE	770	
Manganese	CA-6195-AA-D	119	
Zinc	CA-79-S-AE	43	
Titanium	6449 - AA	269	
Chromium	CA-79-S-AE	152	
Nickel	162 - AZ	62	

g) Calculation of the value "K"

Using the values of μ_m and density as determined, the value of "K" for equation (2) was calculated relative to each elemental S_{max} value secured from the curves of intensity versus surface finish. The following "K" values were obtained:

Element	"K" (Equation (2))
Silicon	41	-
Iron	44	
Copper	34	
Magnesium	41	
Manganese	38	
Zinc	34	
Titanium	36	
Chromium	28	
Nickel	30	

Average value 36 ± 5

It is difficult to explain the unusually low values for chromium and nickel. One could assume, for chromium, that the straight line section, representing the constant intensity zone, depends for its location only three points which themselves are in rather poor agreement.

The value of "K" at 36 is, as anticipated, higher than the value of 30 found by Jenkins and

Hurley (8), this on the basis of the difference in instrumental geometry. A comparison of the values for $S_{\rm max}$ calculated using K=36 with those from the graphs yields the following, and allows some estimation of the accuracy of expressing $S_{\rm max}$.

Element	S _{max} Calculated		S _{max} Graph
Silicon	60	(6 x 10)	75
Iron	210	(21 x 10)	265
Copper	250	(25 x 10)	235
Magnesium	100	(10 x 10)	115
Manganese	190	(19 x 10)	200
Zinc	260	(26 x 10)	250
Titanium	140	(14×10)	145
Chromium	170	(17×10)	135
Nickel	230	(23 x 10)	195

Thus the use of equation (2) with $K = 36\pm 5$ in the calculation of S_{max} for elements other than listed here should be quite feasible.

If one considers that silicon is the critical element to be determined, relative to its sensivity

to surface finish, and if one considers that a surface finish of 25 μ would be satisfactory for all of the elements investigated, it is possible to secure an adequate surface finish by lathe turning at a final cut of 0.001 inch and a traverse speed of 0.001 inch/revolution. It is understood that the tool profile should be 60° under these conditions.

4. CONCLUSIONS

- a) The investigation indicates the determination of $S_{\rm max}$ for nine elements commonly determined quantitatively. The maximum allowable surface roughness for each element is indicated so that, where only certain elements must be determined, the optimum lathe settings to secure this final finish most rapidly could be ascertained.
- b) The value of "K" is established for the Picker Nuclear Spectrodiffractometer. While this value is not established with high accuracy, because of both the graphic method used and the low values obtained for "K" for chromium and nickel, it allows calculation of S_{max} for other metallic matrices with a sufficient level of accuracy. This is particularly so since ordinarly the S_{max} calculated for any element would suggest an approach to a surface finish twice as fine (e.g. $S_{max} = 200$: lathe turn for $S_{max} = 100$).

c) The method of lathe turning is shown to be capable of the preparation rapidly, and with a satisfactory surface finish, of aluminium alloy samples for analysis by XRF. It should be pointed out that, in its proper application, an initial cut should be taken on the lathe, followed by a final cut at the proper depth and traverse settings.

5. EEFERENCE

- 1. R.E.Michaelis and B.A.Kilday, Advance in X-ray Analysis, 5, Pitman, New York, (1961), 405-411.
- 2. D.S.Flikkema and R.V.Schablaske, Advance in X-ray Analysis, 1, Fitman, New York, (1957), 387-398.
- 3. J.A.F.Gidley, Proceedings of the Buxton Symposium, Hilger and Watts, London (1962),3-11.
- 4. K.M.Bills, J.Inst.Metals, 90,(1962),273-274.
- 5. K.M.Bills, Proceedings of the Buxton Symposium, Hilger and Watts, London (1962),16-24.
- 6. B.A.Kilday and R.E.Michaelis, Appl.Spec., <u>16</u>, (1962),136-138.
- 7. F.Lihl and J.Fishhuber, Z.F.Metallkunde, <u>53</u>, (1962),186-190.
- 8. R. Jenkins and P.W. Hurley, S&AE Bulletin, No.FS1, Philips, Eindhoven, (1966).
- 9. E.I.Gunn, Advance in X-ray Analysis, 4, Fitman, New York, (1960), 382-400.

6. APPENDIX

TABLES AND FIGURES.

TABLE 1

SELECTION OF STANDARD SAMPLES FOR INVESTIGATION

Element nvestigated	Silicon Copper Manganese Iron	Zinc Chromium Magnesium	Titanium	Nickel
Element Pb Al Investigated	83.70	90•20	04.6 94.6 94.50	82.90
超			94.0	
Cr Ni				2.50
전		0.20		
ION			0	
COMPOSITION Zn	06•0	4.70		
COM	0.42	0.22	0.89 0.15	
Mg	1.09	3.34	68.0	1.09
Cu	0.70 3.42 1 identical values	0.29 0.65	0.32	64.0
Si Fe Cu	9.50 0.70 3.42 identical va	0.29	0.65 0.25 0.32	12.10 0.41 0.79
Si	9.50		0.65	12,10
Sample	CA-6195-AA-D CA-6195-AC	CA-79-S-AE	6449 - AA	162-AZ

Where no values are shown for elemental composition, these were low enough to be considered insignificant.

Analysis results as reported by Alcan Laboratories

Table no: 2

LATHE SETTING FOR VARIOUS SURFACE FINISHES

Set	Depth of Cut (ins.)	Traverse Speed (in. per rev.)
0.028	0.028	0.032
0.024	0.024	0.028
0.020	0.020	0.023
0.016	0.016	0.018
0.012	0.012	0.014
0.008	0.008	0.009
0.006	0.006	0.007
0.004	0.004	0.005
0.002	0.002	0.002
0.001	0.001	0.001

Table no: 3

DETERMINATION OF SURFACE FINISH - SET 0.028

Measurement	Scale (div.)	mm	đ	d ²
1	65.4	0.654	0.050	0.002500
2	68.6	0.686	0.018	0.000324
3	74.8	0.748	0.044	0.001936
4	71.4	0.714	0.010	0.000100
5	70.4	0.704	0.000	0.000000
6	72.8	0.728	0.024	0.000576
7	69.9	0.699	0.005	0.000025
8	69.3	0.693	0.011	0.000121
9	72.8	0.728	0.024	0.000576
10	68.2	0.682	0.022	0.000484

Average mm = 0.704

Standard deviation (s) = 0.027

Table no: 4

DETERMINATION OF SURFACE FINISH - SET 0.024

Measurement	Scale (div.)	mm	đ	d ²
1	64.3	0.643	0.050	0.002500
2	62.8	0.628	0.035	0.001225
3	57•4	0.574	0.019	0.000361
4	57•5	0.575	0.018	0.000342
5	56.9	0.569	0.024	0.000576
6	57•4	0.574	0.019	0.000361
7	60.2	0.602	0.009	0.000081
8	59•7	0.597	0.004	0.000016
9	56.9	0.569	0.024	0.000576
10	59•5	0.595	0.003	0.000009

Average mm = 0.593

Table no: 5

DETERMINATION OF SURFACE FINISH - SET 0.020

Measurement	Scale (div.)	mm	đ	d ²
1	52•9	0.529	0.039	0.001521
2	46.8	0.468	0.022	0.000484
3	47.7	0.477	0.013	0.000169
4	48.8	0.488	0.002	0.000004
5	50.4	0.504	0.014	0.000196
6	46.7	0.467	0.023	0.000529
7	49.8	0.498	0.008	0.000064
8	51.0	0.510	0.020	0.000400
9	48.8	0.488	0.002	0.000004
10	46.6	0.466	0.024	0.000576

Average mm = 0.490

DETERMINATION OF SURFACE FINISH - SET 0.016

Measurement	Scale (di	.v.) mm	đ	d ²
1	38.9	0.389	0.009	0.000081
2	38.8	0.388	0.008	0.000064
3	39•5	0.395	0.015	0.000225
4	38.9	0.389	0.009	0.000081
5	40.6	0.406	0.026	0.000676
6	38.0	0.380	0.000	0.000000
7	39•4	0.394	0.014	0.000196
8	39•9	0.399	0.019	0.000361
9	37•7	0.377	0.003	0.000009
10	38.5	0.385	0.005	0.000025

Average mm = 0.380

DETERMINATION OF SURFACE FINISH - SET 0.012

Measurement	Scale (div.)	mm	đ	d ²
1	31.2	0.312	0.024	0.000576
2	28.4	0.284	0.004	0.000016
3	29.2	0.292	0.004	0.000016
4	28.8	0.288	0.000	0.000000
5	27.5	0.275	0.013	0.000169
6	28.4	0.284	0.004	0.000016
7	29.8	0.298	0.010	0.000100
8	30.0	0.300	0.012	0.000144
9	27.8	0.278	0.010	0.000100
10	26.9	0.269	0.019	0.000361

Average mm = 0.288

DETERMINATION OF SURFACE FINISH - SET 0.008

Measurement	Scale (div.)	mm	d.	d ²
.1	20.8	. 0.208	0.014	.0.000196
2	19.1	0.191	0.003	0.000009
3	18.9	0.189	0.005	0.000025
4	19.5	0.195	0.001	0.000001
5	19.9	0.199	0.005	0.000025
6	19.0	0.190	0.004	0.000016
7	18.7	0.187	0.007	0.000049
8	20.1	0.201	0.007	0.000049
9	18.5	0.185	0.009	0.000081
10	19.2	0.192	0.002	0.000004

Average mm = 0.194

DETERMINATION OF SURFACE FINISH - SET 0.006

Measurement	Scale (div.)	mm	đ	d ²
1	15.9	0.159	0.013	0.000169
2	16.1	0.161	0.015	0.000225
3	14.0	0.140	0.006	0.000036
4	14.1	0.141	0.005	0.000025
5	14.0	0.140	0.006	0.000036
6	13.8	0.138	0.008	0.000064
7	14.9	0.149	0.003	0.000009
8	15.0	0.150	0.004	0.000016
9	14.2	0.142	0.004	0.000016
10	14.0	0.140	0.006	0.000036

Average mm = 0.146

DETERMINATION OF SURFACE FINISH - SET 0.004

Measurement	Scale	(div.) mm	đ	d ²
1	10.9	0.109	0.014	0.000196
2	10.1	0.101	0.006	0.000036
3	8.9	0.089	0.006	0.000036
4	8.7	0.087	0.008	0.000064
5	9.4	0.094	0.001	0.000001
6	9.8	0.098	0.003	0.000009
7	9.2	0.092	0.003	0.000009
8	9.6	0.096	0.001	0.000001
9	8.9	0.089	0.006	0.000036
10	9.3	0.093	0.002	0.000004

Average mm = 0.095

DETERMINATION OF SURFACE FINISH - SET 0.002

Measurement	Scale (div.)	mm	đ.	d ²
1	5.9	0.059	0.011	0.000121
2	6 .1	0.061	0.013	0.000169
3	4.0	0.040	0.008	0.000064
4	3.9	0.039	0.009	0.000081
5	4.5	0.045	0.003	0.000009
6	4.8	0.048	0.000	0.00000
7	4.4	0.044	0.004	0.000016
8	4.6	0.046	0.002	0.000004
9	4.8	0.048	0.000	0.000000
10	4.9	0.049	0.001	0.000001

Average mm = 0.048

Table no: 12

DETERMINATION OF SURFACE FINISH - SET 0.001

Measurement	Scale (di	v.) mm	d	d ²
1	2.9	0.029	0.004	0.000016
2	3 . 1	0.031	0.006	0.000036
3	2.4	0.024	0.001	0.000001
4	2.1	0.021	0.004	0.000016
5	2.0	0.020	0.005	0.000025
6	2.6	0.026	0.001	0.000001
7	2.4	0.024	0.001	0.000001
8	2.8	0.028	0.003	0.000009
9	2.1	0.021	0.004	0.000016
10	2.5	0.025	0.000	0.00000

Average mm = 0.025

Table no: 13

DETERM	INATION OF SUR	FACE FINISH	- SET	<u>6 μ</u>
Measurement	Surfmeter reading (µ i	ms.)	đ	d ²
1	7	0.178	0.025	0.000625
2	7	0.178	0.025	0.000625
3	8	0.203	0.000	0.000000
4	8	0.203	0.000	0.000000
5	8	0.203	0.000	0.000000
6	9	0.229	0.026	0.000676
7	9	0.229	0.026	0.000676
8	8	0.203	0.000	0.000000
9	8	0.203	0.000	0.00000
10	7	0.178	0.025	0.000625

Average $\mu = 0.203$

Table no: 14

SURFACE FINISH IN MICRONS FOR EACH TEST SET

Set	I	nm	μ	Value for graphing
0.028	0.70	± 0.03	(70 ± 3)x10	700
0.024	0.59	± 0.03	$(59 \pm 3) \times 10$	600
0.020	0.49		(49 ± 2)x10	500
0.016	0.38		$(38 \pm 1) \times 10$	400
0.012	0.29	± 0.01	$(29 \pm 1) \times 10$	300
0.008	0.194	± 0.007	194 ± 7	200
0.006	0.146	± 0.008	146 ± 8	150
0.004	0.095	± 0.007	95 ± 7	100
0.002	0.048	± 0.007	48 ± 7	50
0.001	0.025	± 0.004	25 ± 4	25
Diamond paste 6	μ	-	0.20 = 0.02	0

Table no: 15

SILICON CARBIDE ABRASIVE PAPER SURFACE FINISHES

(measured with Surfmeter)

Grit class	Microinches µ (10 ⁻⁶ in.)	Microns μ (10 ⁻⁶ m)
80	85	2.1
120	50	1.3
240	40	1.0
320	28	0.7
400	20	0.5
600	8	0.2
Diamond paste 6 μ	8	0.2

TABLE 16

X-RAY SPECTROMETER PARAMETERS FOR SURFACE FINISH STUDY

	Cu	Cr	F.	иM	ŗN	Ţ	Zn	Mg	Si
Target	Μ	M	M	M	M	М	M	占	Cr
Kv	07	777	07	70	70	04	04	50	50
mA	50	56	36	50	77	30	18	32	30
Crystal	H H	LiF	LiF	Lif	LiF	LiF	LiF	ADP	PET
w angle	+ 0.19	+ 0.19	+ 0.19	+ 0.19	+ 0.19	+ 0.19	+ 0.19	44.0 -	- 0.47
Collimator	fine	coarse	fine	coarse	fine	coarse	fine	coarse	coarse
Counter	SG	SG	SC	SC	SC	PFC	SC	PFC	PFC
He flow			1.5 c.	cfh for all	examinations	នព			
P-10 flow	ī	ı	ı	I	ı	0.5 cfh	ı	0.5 cfh	0.5 cfh
Gain	2	70	70	2	7.7	70	70	72	5
Counter HV	832	892	832	880	832	1520	832	1530	1534
Lower level kev 0.50	os 0.50	0.50	0.50	0.50	0.50	3.72	0.50	0.50	98.0
Upper level kev zero	y ze r o	zero	zero	zero	zero	2.00	zero	2.00	2.62
Mod e	Int	Int	Int	Int	Int	700 %	Int	100 %	100 %
Counting Secs	10	10	10	10	10	10	10	10	10
2 theta peak	45.02	69.34	57.52	62.94	99•87	86.12	41.78	136.69	109.21
Background 1	44.30	00•89	56.52	61.50	47.50	84.40	41.20	135.40	107.00
Background 2	45.80	70.50	58.52	94.50	79.50	88	42.50	138.50	111.50
Sample rotation	ц			All sa	samples ${f r}$ otated	ted			

Counting standard:-		CA-6195-AA-D	Standard	ard cour	ıt taken	count taken for set 0.024	II	1567 cps				
Test sample:-	CA-61	CA-6195-AC										
Set Set count	Average cps	Correction Factor	Peak	Bkg 1	Bkg 2 cps	Average peak	Average bkg	Corrections to Average Peak Dead time av. bkg. Factor	av. bkg.	ge Peak Factor	Diff	38
0.024 1590,1557, 1555	1567	1.0000	962,957, 16, 959	16, 14,15	7,8,7	656	디	ı	87/6	876	+ 95	28
0.020 1580,1549, 1552	1560	1.0045	1052,1062, 1043	17, 17, 17, 17,	6,86	1052	13	ı	1039	1043	6 6	31
0.016 1581,1547, 1545	1558	1.0058	1144,1136,	18,	9,9,	1138	13	ı	1125	1131		34
0.012 same as 0.016	1558	1.0058	1234,1222, 18, 1231	18, 18, 18, 18, 18, 18, 18, 18, 18, 18	6,6	1229	13	ı	1216	1223	+ -	37
0.008 1575,1562, 1559	1565	1,0013	1372,1356, 18, 1358	18,	10,9,	1362	77	ı	1348	1350	7 7	32
0.006 same as 0.008	1565	1,0013	1430,1411, 19, 1402 19,18	19,11	10,10,	1414	174	ı	1400	7402	+ +	35
0.004 1598,1575,	1581	0.9911	1529, 1491, 19, 1490	., 19, 18,20	10,11,	1503	77	ı	1489	9741		36
0.002 same as 0.004	1581	0.9911	1568,1547, 19, 1555	, 19, 20,19	10,11, 10	1557	77	ı	1543	1529	+ 53	38
0.001 1531,1532,	1542	1.0162	1540,1501, 21, 1556 20,22	20,22	10,11, 10	1532	15	ı	1517	1541		39
6 u same as	1542	1.0162	1.0162 1540,1522, 22, 1518 21,20	22,	10,12, 12	1527	16	ı	1511	1535	o I	38

INTENSITY OF EMISSION VERSUS SURFACE FINISH

SILLCON

TABLE -7

FINISH
SURFACE
छ
OF EMISSION VERSI
INTENSITY

Standard count taken for set 0.016 = 4014 cps

Counting standard:- 143-AA

IRON

	0							-					
Test	Test sample:-	CA-6195-AC	5-AC										
Set	Set count	Average cps	Correction Factor	Peak]	Bkg l	Bkg 2 A	Av er age peak	Ave r age bkg	Corrections to Dead time av.	s to Averagave bkg.	Average Peak bkg. Factor	Diff	38
0.028	4062,4028, 3988	4026	0.9970	4708,4681, 4686	, 358, 1 350,361 1	109,107, 106	7695	231	ı	4461	4448	Ø	29
0°024	0.024 4010,3992, 3985	3996	1.0045	5288,5250, 5238,33	357,363	112,107, 114	5259	236	ı	5023	5046	067+	99
0.020	0.020 same as 0.024	3996	1.0045	5864,5923, 368, 5838 364,371		116,112, 117	5875	242	l	5633	5658	+ + + + + + + + + + + + + + + + + + +	70
0.016	0.016 3974,4046, 4021	4014	1,0000	6498,6479, 376, 6489, 372,378		118,115, 119	2849	246	ı	6241	6241	(96)	77
0.012	0.012 same as 0.016	4014	1,0000	6849,6815, 378, 6828 372,381		120,118, 118	6830	24,8	I	6582	6582	+241 -03	75
0,008	0.008 4271,4307, 4324	4301	0.9333	7431,7490, 453, 7406 450,456		122 , 129 , 129	7442	290	ı	7152	9299	C4 - 17	92
900•0	same as	4301	0.9333	7539,7476, 7453,454	461,	128,124, 121	7489	290	ı	7199	6719	† C	77
700°0	0.004 4015,3998, 4009	4004	1.0017	7036,6980, 6966	0, 468, 1 472,470 1	118,120, 124	7669	296	ı	8699	6029) T	92
0.002	0.002 same as 0.004	4004	1.0017	7072,7000, 6964 4,	0, 480, 1 474,476	118,116, 1114	7012	297	I	6715	6726	~ α 	77
0.001	0.001 4013,3993, 3978	3995	1.0048	7043,7000, 6978 4,	474, 1 472,471	116,118, 118	7007	295	ı	6712	4479	-17	77
n 9	same as	3995	1,0048	7043,6880, 470, 7056 477,472	470, 1 17,472	116,116,	6669	295	ı	8699	6730		77

SITY OF EMISSION VERSUS SURFACE FINISH	
INTENSITY OF	
COPPER	

Counting standard:- 143-AA Standard count taken for set 0.016 = 28407 cps

CA-6195-AC Test sample:-

	Diff 3s	149	+1447	+14,50 166	+1408	+1283 164	+793	-1 33 167	168	168	169	(
		24,821	47	+1 27698	29106	30389	47	31049	+79 31128	+31	+85 31244	((
	verage kg• Fa											
-	Av. Bkg.	24890	26268	27698	29106	30389	29672	29545	31004	31035	31085	
-	Corrections to Average Feak Dead time Av. Bkg. Factor	I	I	ı	1	I	ı	1	ı	ı	l	
•	Average bkg	228	241	251	251	248	277	283	271	277	276	
•	Average peak	25118	26509	27949	29357	30637	29949	29828	31275	31312	31361	
ī	BKg 2 cps	209,212, 207	230 , 225 224	241,238 237	2 35, 232 , 230	230,228 233	248,243 237	260,250 242	241,239 247	248,245 241	242 , 239 242	
į	BKg 1 cps	58 248, 252,240	14 255 261 , 249	48 270 255,262	56 268 264 , 277	23 270 266 , 260	13 306 316 , 312	54 310 324,309	08 301 304,298)2 315 310 , 303	03 315 310 , 306	
f	Cps	25210,25058 25086 25	26623,26514 255 26390 261,249	27999,27848 270 28001 255,262	29430,29366 29276	30 6 95,30503 270 30713 266,260	29995,29913 306 29941 316,312	29994,29764 29726	31303,31208 31314,30	31372,31202 315 31362 310,303	31429,31303 31351	
-	Correction (0.9972	1,0000	1,0000	1,0000	1.0000	1.0509	1.0509	1.0040	1.0040	1.0051	
	Av er age cps	28487	28414	28414	7 28407	28407	27031	27031	5 28293	28293	28263	
-	Set count	28580 , 28430 28487 28451	28489,28352 28414 28401	same as 0.024	0.016 28353,28407 28407 28460	0.012 same as 0.016	0.008 27127,27002 27031 26964	same as 0.008	28280,28396 28204	0.002 same as 0.004	0.001 28273,28310 28263 28206	
	Set	0.028	0.024	0.020	910.0	0.012	800.0	900•0	700.0	0.002	0.001	

MAGNESIUM

INTENSITY OF EMISSION VERSUS SURFACE FINISH

Counting standard:-

340-AA-D

Standard count taken for set 0.016 = 533 cps

Test sample:-

CA-79-S-AE

Set	Set count cps	Average cps	Correction Factor	Peak cps	Bkg l cps	Bkg 2 cps	Average peak	Average bkg	Correction Dead time	s to Avera Av. Bkg.	ge Peak <u>Factor</u>	Diff	3s
0.024	584,571,568	574	0.9286	71,59,62	20,17,19	10,9,11	64	15	-	49	45	+19	6
0.020	same as 0.024	574	0.9286	87,80,80	18,19,19	9,9,9	82	114	-	68	64		6
0.016	565,528,505	533	1.0000	107,99,102	2 19,18,16	9,9,8	102	13	-	89	89	+25 +26	8
0.012	same as 0.01	.6 533	1.0000	136,127, 125	17,19,19	11,9,12	129	14	-	115	115		10
0.008	506,519,502	509	1.0471	139 , 139 , 139	18,19,15	10,7,9	139	11	-	128	134		11
0.006	same as 0.00	8 509	1.0471	165 , 155 , 158	19,20,15	8,10,9	159	13	-	146	152		11
0.004	508,490,506	501	1.0658	167,165, 158	14,15,16	8,8,7	163	12	-	151	161		11
0.002	same as 0.00	04 50l	1.0658	168,174, 168	15,16,14	9,9,8	170	12	-	158	168		11
0.001	510,485,509	489	1.0702	169,168, 158	14,13,15	8,7,7	165	10	-	155	166		11
6 u	same as 0.00	1 489	1.0702	167,160, 171	14,16,13	8,7,9	166	11	-	155	165	- 1	11

MANGANESE

INTENSITY OF EMISSION VERSUS SURFACE FINISH

Counting standard:-

143**-**AA

Standard count taken for set 0.016 = 2403 cps

Test sample:-

CA-6195-AC

1050 5	campito.	011	22/2 110										
Set	Set count cps	Average cps	Correction Factor	Peak cps	Bkg l cps	Bkg 2 cps	Average peak	Average bkg	Correction Dead time	s to Avera Av. Bkg.	ge Peak <u>Factor</u>	Diff	3s
0.028	2401,2 3 92, 2418	2404	1.0000	2529,2540, 2540	161,158, 154	129 , 131 , 130	2536	144	-	2392	2392		45
0.024	2420,2418, 2401	2413	0.9959	2765,2733, 2741	164,160, 159	133,130, 129	2746	146	-	2600	2589	+197	49
0.020	same as 0.024	2413	0.9959	3038,2978, 2961	170,174, 168	136,130, 134	2992	171	-	2840	2828	+239	48
0.016	2387,2416, 2406	2403	1.0000	3194,3125, 3147	170,177, 170	138 , 138 , 139	3155	155	-	3000	3000	+172	51
0.012	same as 0.016	2403	1.0000	3403,3323, 3351	174,170, 170	137 , 136,	3361	154	-	3207	3207	+207	54
0.008	2514,2512, 2513	2513	0.9563	3703,3629, 3647	175,183, 181	155,148, 148	3660	165	-	3495	3342	+135	56
0.006	same as 0.008	2513	0.9563	3712,3685, 3685	181,178, 184	153,149, 156	3694	167	-	3527	3373	+31	57
0.004	2356,2348, 2360	2355	1.0204	3498,3482, 3503	175,170, 169	138,140,	3494	154	-	3340	3408	+35	58
0.002	same as 0.004	2355	1.0204	3537,3503,	174,169,	135,140,	3517	154	-	3363	3432	+24	58
0.001	2348,2364,	2353	1.0212	3511 3499,3470,	169,174,	136	3491	153	-	3338	3409	- 23	58
6 u	2348 same as 0.001	2353	1.0212	3504 3520,3511, 3492	169 171,170, 166	133 136,141, 139	3508	154	-	3354	3425	+16	58

ZINC INTENSITY OF EMISSION VERSUS SURFACE FINISH

Counting standard:-

6561-CABS

Standard count taken for set 0.016 = 24108 cps

Test sample:-

CA-79-S-AE

Set	Set count cps	Average cps	Correction Factor	n Peak cps	Bkg l cps	Bkg 2 cps	Average peak	Average bkg	Correction Dead time	s to Avera	ge Peak Factor	Diff	3s
0.028	23845,23781, 23772	23799	1.0130	35876,35799 35719	, 362 350,344	336,350, 316	35798	340	35845	35505	35963	7.42.50	187
0.024	same as 0.028	3 23799	1.0130	37552 , 37448 37423	, 426 401,398	372,361, 346	37474	384	37680	37296	37780	+1817	185
0.020	24285,24345, 24039	24223	0.9953	41594 , 41474 41536	, 369 375 , 379	323 ,3 23 338	41534	351	41899	41548	41351	+3571	203
0.016	24153,24098, 24074	24108	1.0000	43725 , 43621 43660	, 382 385,395	341 , 336 343	43669	364	44129	43765	43765	+2414	215
0.012	same as 0.016	24108	1.0000	46152,46052 46972	414 412 , 403	359 , 358 346	46059	382	46634	46252	46252	+2487	194
0.008	24 3 69,24212, 24287	24289	0.9926	46652,46739 46633	636 635 , 650	403,401 410	46675	524	47281	46757	46409	+157	195
0.006	same as 0.008	3 24289	0.9926	47025 , 46940 46987	643 641 , 650	418,404 403	46984	527	47606	47079	46728	+319	196
0.004	24632,24600, 24614	24615	0.9794	47486,47419 47411	449 438 , 435	385 , 371 370	47439	408	48085	47677	46695	- 33	196
0.002	same as 0.001	24615	0.9794	47608 , 47512 47538	482 469 , 472	398 , 402 400	47553	437	48204	47767	46783	+88	197
0.001	24534,24633, 24601	24589	0.9804	47618 , 47520 47567		400,385	47568	419	48409	47990	47051	+268	198
6 u	same as 0.001	24589	0.9804	47563,47482		375,370	47544	423	48195	47772	46838	-213	194

TITANIUM

INTENSITY OF EMISSION VERSUS SURFACE FINISH

Test sample:-	Counting standard:-
6449-AA	6046-CABS
	Standard count taken for set $0.016 = 259$ cps

6 u	0.001	0.002	0.004	0.006	0.008	0.012	0.016	0.020	0.024	0.028	Set
same as 0.001	272,264,278	same as 0.004	268,275,270	same as 0.008	261,269,259	same as 0.016	256,264,256	same as 0.024	255,26 2 ,260	258,269,256	Set count
271	271	271	271	263	263	259	259	259	259	261	Av era ge cps
0.956	0.956	0.956	0.956	0.985	0.985	1.000	1.000	1.000	1.000	0.992	Correction Factor
272,264,272 3,3,2	268,260,264 3,3,2	270,264,272 3,3,3	259,252,254	240,239,238 3,3,2	212,219,213 3,3,2	195,196,188, 2,2,1	178,170,167 2,2,1	149,146,141	126,121,126 2,2,1	105,109,107	Peak cps
3,3,2	3,3,2	3,3,3	2,2,3	3,3,2	3,3,2	2,2,1	2,2,1	2,2,2	2,2,1	1,1,2	Bkg 1
3,2,2	3,3,3	3,3,2	3,3,2	2,2,3	2,3,2	1,2,2	1,2,2	2,2,1	2,1,1	1,2,1	Bkg 2
269	264	265	255	239	215	193	170	145	124	107	Average peak
			N	N	N	N	N	N	N	Ь	Averag bkg
I	1	1	1	ı	I	ı	1	1	1	I	e Corrections to Average Peak Diff Dead time Av. Bkg. Factor
			253	237	213	191	168	143	122	106	s to Avera
255	249	251	242	233	210	191	168	143	122	105	ge Peak Factor
d	<u> </u>	5 1	+0	t &	- 1 33 F	10 6	3 8	- 25 - 25 - 25 - 25 - 25 - 25 - 25 - 25	- + -> -	777	Diff
17	17	17	17	16	15	14	12	L	· 1	10	38

TAB. 24

CHROMIUM

INTENSITY OF EMISSION VERSUS SURFACE FINISH

Counting standard:-143-AA Standard count taken for set 0.016 = 686 cps

Test sample:-

CA-79-S-AE

6 u	0.001	0.002	0.004	0.006	0.008	0.012	0.016	0.020	0.024	0.028	Set
same as 0.001	1 680,688,678	2 same as 0.004	4 714,710,705	6 same as 0.008	8 707,720,704	2 same as 0.016	6 680,702,677	0 s ame as 0.024	4 700,689,675	8 705,691,680	Set count
682	682	710	710	710	710	686	686	688	688	692	Ave ra ge
1.0060	1.0060	0.9662	0.9662	0.9662	0.9662	1.0000	1.0000	0.9971	0.9971	0.9913	Correction Factor
2217 , 2219 , 2212	2198,2190, 2206	2291,2306, 2295	2272,2275, 2286	2245,2231, 2241	2109,2118, 2097	1925 , 1881 , 1894	1744,1742, 1785	1639,1618, 1624	1494,1478, 1488	1347 , 1331 , 1336	Peak cps
	198,200, 196				215 , 214, 211	201,204. 207	194 , 194. 192		196 , 194, 189		Bkg 1
210,199, 189,181, 200 176	, 188,187, 178	205,200, 190,179, 197	205,206, 190,193, 200 191	210,212, 181,192, 213	215,214, 194,194, 211	201,204, 180,187, 207 180	194,194, 177,175, 192	194,189, 177,185, 181 181	196,194, 180,179, 189	191,190, 175,171, 187	Bkg 2
2216	2198	2297	2278	2239	2108	1900	1757	1627	1487	1338	Ave r age peak
192	191	193	198	199	202	193	185	185	186	181	Average bkg
I	l	l	ı	ı	ı	l	1	ı	l	ı	Corrections to Average Peak Dead time Av. Bkg. Factor
2024	2007	2104	2080	2040	1906	1707	1572	1442	1301	1157	s to Avers
2036	2019	2033	2001	1971	1842	1707	1572	1438	1297	1147	ge Peak Factor
	+17	-14	+32	+30	+129	+135	+135	+134	+141	+150	Diff
43	43	43	42	41	39	43	39	36	32	34	ည္တ

NICKEL

INTENSITY OF EMISSION VERSUS SURFACE FINISH

Counting standard:-

6046-CABS

Standard count taken on set 0.016 = 27426 cps

Test sample:-

162**-**AZ

Set	Set count	Average cps	Correction Factor	Peak cps	Bkg l cps	Bkg 2 cps	Average _Peak_	Average Bkg	Correction Dead_time	s to Avera Av. Bkg.	ge Peak Factor	Diff	3s
0.028	27460,27492, 27431	27461	0.9987	26068,25964 25953	143,140, 144	231,225, 224	25995	184	-	25811	25778		139
0.024	27500,27410, 27430	27447	0.9992	27621,27541 27616	148,144,	229,236, 233	27593	189	-	27404	27382	+1604	148
0.020	same as 0.024	27447	0.9992	29301,29189 29248	149,151, 146	214,208, 218	29246	182	-	29064	29041	+1659	157
0.016	27455,27384, 27438	27426	1.0000	30986,30907 30896	,150,150, 149	216,222, 221	30924	185	30 9 32	30747	30747	+1706	166
0.012	same as 0.016	27426	1.0000	32581 , 32511 32467	165,171, 172	249,250, 254	32520	210	32 5 70	32360	32360	+1613	174
0.008	27366,27403, 27419	27396	1.0011	33588 , 33639 33654	206,207, 194	249,258, 254	33627	227	33708	33481	33518	+1158	174
0.006	same as 0.008	27396	1.0011	33856,33831 33726	196,194, 184	254,246, 245	33804	220	33891	33671	33708	+190	175
0.004	27790,27684, 27664	27713	0.9896	34438 , 34362 34356	,199,201, 200	261,259, 255	34385	229	34476	34247	33891	+183	176
0.002	same as 0.004	27713	0.9896	34198,34141 34153	200,204, 199	258,249, 255	34164	228	34254	34026	33672	- 219	175
0.001	27782,27696, 27665	27714	0.9896	34438,34362 34398	192,198, 203	261,258, 255	34399	228	34491	34263	33907	+235	176
6 u	same as 0.001	27714	0.9896	34341 , 34282 34286		268,260, 262	34303	233	34390	34157	33802	-1 05	176

CALCULATION OF MASS ABSORPTION COEFFICIENT FOR ALLOY MATRIX

Alloy: CA-6195-AA-D

Si-Si

 $\mu \text{Si}(\text{Si}) \times \text{cSi} = 315 \times 0.095 = 30$

<u>Si-Fe</u>

 $\mu \text{Si(Fe)} \times \text{cFe} = 2040 \times 0.007 = 14$

<u>Si-Cu</u>

 μ Si(Cu) x cCu = 2415 x 0.0342 = 83

<u>Si-Mg</u>

 $\mu \text{Si}(Mg) \times cMg = 2660 \times 0.0109 = 29$

<u>Si-Mn</u>

 $\mu \text{Si}(Mn) \times cMn = 1920 \times 0.0042 = 8$

<u>Si-Zn</u>

 $\mu \text{Si}(Zn) \times cZn = 2510 \times 0.009 = 23$

Si-Al

Other elements are low order

MANGANESE $K_{\mathcal{K}} = 2.10 \text{ Å}$

Mn-Mn

μMn(Mn) x cMn = 80 x 0.0042 = 0.3

Mn-Fe

μMn(Fe) x cFe = 91 x 0.007 = 0.6

Mn-Cu

μMn(Cu) x cCu = 123 x 0.0342 = 4.2

Mn-Mg

μMn(Mg) x cMg = 95 x 0.0109 = 1

Mn-Si

μMn(Si) x c Si = 146 x 0.095 = 14

Mn-Zn

μMn(Zn) x cZn = 135 x 0.009 = 1.2

Mn-Al

μMn(Al) x cAl = 117 x 0.837 = 98

Other elements are low order

 μ matrix = 119

119

COPPER $K_{\star} = 1.54 \text{ Å}$

Cu-Cu $\mu Cu(Cu) \times cCu = 55 \times 0.0342 = 2$ Cu-Fe $\mu Cu(Fe) \times cFe = 316 \times 0.007 = 2$ Cu-Mg $\mu Cu(Mg) \times cMg = 40 \times 0.0109 = 0.4$ Cu-Mn $\mu Cu(Mn) \times cMn = 287 \times 0.0042 = 1.2$ Cu-Si $\mu Cu(Si) \times cSi = 63 \times 0.095 = 6$ Cu-Zn $\mu Cu(Zn) \times cZn = 60 \times 0.009 = 0.5$ Cu-Al $\mu Cu(Al) \times cAl = 45 \times 0.837 =$ 37.7 50

Other elements are low order

Fe-Fe

μFe(Fe) x cFe = 71 x 0.007 = 0.5

Fe-Cu

μFe(Cu) x cCu = 96 x 0.0342 = 3.3

Fe-Mg

μFe(Mg) x cMg = 77 x 0.0109 = 0.8

Fe-Mn

μFe(Mn) x cMn = 64 x 0.0042 = 0.3

Fe-Si

μFe(Si) x cSi = 116 x 0.095 = 1.1

Fe-Zn

μFe(Zn) x cZn = 110 x 0.009 = 1

 $\mu Fe(Al) \times cAl = 94 \times 0.837 =$

Other elements are low order

 μ matrix = 86

86

Alloy: CA-79-S-AE

ZINC $K_{\star} = 1.44 \text{ Å}$

Zn-Zn

 $\mu Zn(Zn) \times cZn = 49 \times 0.047 = 2.3$

Zn-Cu

 $\mu Zn(Cu) \times cCu = 45 \times 0.0065 = 0.3$

Zn-Fe

 $\mu Zn(Fe) \times cFe = 261 \times 0.0029 = 0.8$

Zn-Mg

 $\mu Zn(Mg) \times cMg = 35 \times 0.0334 = 1.2$

Zn-Mn

 $\mu Zn(Mn) \times cMn = 238 \times 0.0022 = 0.5$

Zn-Cr

 $\mu Zn(Cr) \times cCr = 200 \times 0.002 = 0.4$

Zn-Al

 $\mu Zn(Al) \times cAl = 41 \times 0.902 = 37$

Other elements are low order

CHROMIUM $K_{\star} = 2.29 \text{ Å}$

Other elements are low order

MAGNESIUM $K_{\star} = 9.89 \text{ Å}$

.Mg-Mg				
anne de la companya d	μMg(Mg)	x cMg =	350 x 0.034 =	12
Mg-Si	μMg(Si)	x cSi =	740 x 0.0018 =	1
Mg-Cu Mg-Fe	μMg(Cu)	x cCu =	5035 x 0.0065 =	33
	μ Mg(Fe)	x cFe =	4100 x 0.0029 =	12
Mg-Mn Mg-Zn	μMg(Mn)	x cMn =	3790 x 0.0022 =	8
***************************************	μMg(Zn)	x cZn =	5235 x 0.047 =	246
Mg-Cr	μMg(Cr)	x cCr =	3510 x 0.002 =	7
	μMg(Al)	x cAl =	500 x 0.902 =	451
				770

Other elements are low order

Alloy: 6449-AA

TITANIUM $K_{\infty} = 2.75 \text{ Å}$

Ti-Ti $\mu \text{Ti}(\text{Ti}) \times \text{cTi} = 114 \times 0.0008 = 0.1$ Ti-Cu $\mu \text{Ti}(\text{Cu}) \times \text{cCu} = 262 \times 0.0032 = 0.8$ Ti-Fe $\mu \text{Ti}(\text{Fe}) \times c \text{Fe} = 193 \times 0.0025 = 0.5$ Ti-Mg $\mu \text{Ti}(Mg) \times cMg = 212 \times 0.0089 =$ 2 Ti-Mn $\mu \text{Ti}(Mn) \times cMn = 171 \times 0.0015 = 0.3$ Ti-Si $\mu \text{Ti}(\text{Si}) \times \text{cSi} = 328 \times 0.0065 = 2$ Ti-Bi $\mu \text{Ti}(\text{Bi}) \times \text{cBi} = 970 \times 0.0046 = 4.5$ Ti-Pb $\mu \text{Ti}(Pb) \times cPb = 935 \times 0.0046 = 4.3$ Ti-Al $\mu \text{Ti}(Al) \times cAl = 263 \times 0.965 = 254$ 269

Other elements are low order

Alloy: 162-AZ

NICKEL $K_{\chi} = 1.66 \text{ Å}$

Ni-Ni

 $\mu \text{Ni(Ni)} \times \text{cNi} = 61 \times 0.025 = 2$

Ni-Cu

 $\mu \text{Ni}(\text{Cu}) \times \text{cCu} = 65 \times 0.0079 = 0.5$

Ni-Fe

 $\mu \text{Ni(Fe)} \times cFe = 397 \times 0.0041 = 2$

<u>Ni-Ng</u>

 $\mu \text{Ni(Mg)} \times \text{cMg} = 48 \times 0.0109 = 0.5$

<u>Ni- Si</u>

 $\mu \text{Ni}(\text{Si}) \times c \text{Si} = 76 \times 0.121 = 9$

Ni-Al

 $\mu \text{Ni(Al)} \times \text{cAl} = 58 \times 0.829 = 48$

Other elements are low order

μ matrix = 62

